

## TWO TYPES OF ELECTROGENIC, H<sup>+</sup>-PUMPING ATPases IN MEMBRANE VESICLES FROM SOYBEAN ROOTS

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(Received October 19, 1985; Accepted November 18, 1985)

### Abstract

Microsomal membrane vesicles from soybean (*Glycine max* L. cv. Williams) roots were prepared with a 10% dextran cushion and tested for electrogenic H<sup>+</sup>-pumping ATPases. The ATP-dependent uptake of <sup>51</sup>CrCN<sup>-</sup> which could be reversed through the addition of carbonyl cyanide *m*-chlorophenylhydrazone (CCCP) reflected the existence of an ATP-fueled electrogenic H<sup>+</sup> pump which generated positive membrane potentials in these vesicles. Quinacrine fluorescence quenching was induced by MgATP and stimulated by Cl<sup>-</sup>. Fluorescence quenching could be reversed by CCCP or gramicidin (H<sup>+</sup> conductors) and provided evidence that an ATP-fueled pH gradient (inside acid) was formed across the membrane of these vesicles. ATP-dependent pH gradient was partially decreased by ADP or AMP, and completely inhibited by N,N'-dicyclohexylcarbodiimide (DCCD). About 30-50% of the pH gradient generated was sensitive to vanadate and the rest was sensitive to nitrate. The vanadate-sensitive ATPase was enriched at 1.16-1.18 g/ml and showed optimal activity at pH 6.3. The nitrate sensitive ATPase was enriched in vesicles of 1.10-1.12 g/ml and showed pH optimum of 8.1. These properties suggest the vanadate-sensitive and nitrate-sensitive H<sup>+</sup>-pumping ATPases are associated with the plasma membrane and tonoplast, respectively. K<sup>+</sup> dissipated ΔpH generated by either H<sup>+</sup> pump, indicating that a H<sup>+</sup>/K<sup>+</sup> exchange system exists on the plasma membrane as well as the tonoplast.

**Key words:** ATPase; electrogenic H<sup>+</sup>-pump; *Glycine max*; membrane vesicle; quinacrine fluorescence; SCN<sup>-</sup> uptake.

### Introduction

Electrophysiology studies of intact soybean roots have demonstrated the existence of an electrogenic pump across the plasma membrane (Chen and Spanswick, 1982; Lew and Spanswick, 1984a). To identify the electrogenic pump, we have isolated membrane vesicles from soybean roots and tested for the presence of ATP-driven ion pumps.

Electrogenic  $H^+$  pumps have been demonstrated recently in various plant tissues using isolated membrane vesicles (Sze, 1985). We show here that there are two electrogenic,  $H^+$ -pumping ATPases in soybean roots as found in other plant tissues. We also detect a  $H^+/K^+$  exchange system on both the plasma membrane and the tonoplast of soybean roots. Preliminary results of this study have been reported (Chen and Sze, 1984).

### Materials and Methods

#### *Plant Material*

Soybean seeds (*Glycine max* L. cv. Williams) were soaked in an aerated solution of 0.5 mM  $CaSO_4$  for 3 h before incubating in the dark for germination. They were germinated at 25°C between two layers of moistened tissue paper in boxes covered with aluminum foil. After 2 days of growth, the tips (1–1.5 cm) of the roots were harvested.

#### *Preparation of Sealed Microsomal Vesicles*

Sealed microsomal vesicles were prepared as described by Sze (1980) with some modifications. Root tips (5–8 g) were homogenized twice with a mortar and pestle in 35 ml of medium containing 250 mM mannitol, 25 mM HEPES-bis tris propane (BTP), pH 7.4, 3 mM ethylene glycol-bis-( $\beta$ -aminoethyl ether)-N, N'-tetraacetic acid (EGTA), 2 mM dithiothreitol, 0.5% (w/v) bovine serum albumin and 0.5 mM phenylmethylsulfonyl fluoride. After filtration through two layers of Miracloth, the homogenate was centrifuged for 15 min at 13,000 g. The supernatant was centrifuged for 30 min at 60,000 g. The 60,000 g pellet (microsomal fraction) was resuspended in 250 mM mannitol, 2.5 mM HEPES-BTP, pH 7.4 and 1 mM dithiothreitol, layered over a 10% (w/w) dextran cushion prepared in 250 mM mannitol and 2.5 mM HEPES-BTP. After centrifugation for 2 h at 70,000 g, the white interface on top of the dextran cushion was collected and referred as sealed microsomal vesicles.

#### *ATPase Assay*

ATPase activity was measured as described (Hodges and Leonard, 1974) and briefly indicated in the Legend to Table 1. Protein was measured by the method of Lowry *et al.* (1951), after precipitation with 10% (w/v) trichloroacetic acid using bovine serum albumin as the standard.

#### *SCN<sup>-</sup> Uptake*

SCN<sup>-</sup> uptake was measured by a filtration method described by Sze and Churchill (1981) except the reaction mixture containing 16  $\mu$ M  $S^{35}CN^-$ . SCN<sup>-</sup> is a

permeant anion and ATP-dependent S<sup>14</sup>CN<sup>-</sup> (Amersham) uptake indicated the generation of an inside positive membrane potential.

#### Quinacrine Fluorescence Quenching

Quinacrine fluorescence was measured with a Turner model 430 spectrofluorometer by a procedure described by Churchill and Sze (1983) and briefly indicated in the legend to Fig. 2. MgSO<sub>4</sub>, BTP-Cl or ATP was added to a reaction mixture of 25°C to initiate H<sup>+</sup> pumping. The fluorescence of quinacrine is quenched when the amine moves into an acidic compartment (Lee *et al.*, 1982).

#### Chemicals

Sodium ATP was purchased from Boehringer-Mannheim and converted by Dowex ion exchange and bis-tris-propane titration to ATP-BTP at pH 6.8. Most other chemicals were from Sigma.

### Results

#### ATP-generated Membrane Potential

SCN<sup>-</sup> accumulation into the sealed microsomal vesicles was stimulated by ATP and reached a steady state within 5 min (Fig. 1). The enhancement by ATP implied that ATPases on the membrane vesicles generated an inside positive membrane potential and the positive potential provided the energy to drive SCN<sup>-</sup> uptake. SCN<sup>-</sup> accumulation could be quickly decreased by the protonophore CCCP and this suggested that H<sup>+</sup> might be the electrogenic ion.

#### Effects of Ionophores and Inhibitors on ATPase Activity

Protonophore CCCP and ionophore gramicidin stimulated ATPase activities

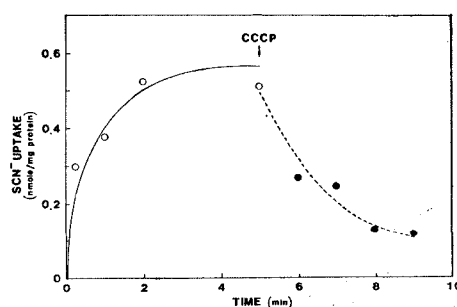


Fig. 1. ATP-induced SCN<sup>-</sup> accumulation into microsomal vesicles. CCCP (5  $\mu$ M) decreased SCN<sup>-</sup> uptake. Reaction mixture consisted of 25 mM HEPES-BTP, pH 7.0, 3 mM MgSO<sub>4</sub>, 16  $\mu$ M [<sup>14</sup>C] SCN, 0.4 mg/ml membrane vesicles with or without 3 mM ATP.

(Table 1) in these sealed microsomal vesicles which agreed with the idea of an ATP-driven electrogenic  $H^+$  pump. ATP hydrolysis is apparently decreased when an electrochemical gradient is generated and dissipation of the  $H^+$  electrochemical gradient by CCCP or gramicidin releases this inhibition.

**Table 1.** *Effect of inhibitors and ionophores on ATPase activity in microsomal vesicles from soybean roots*

ATPase activity was determined by measuring  $P_i$  release. The reaction mixture consisted of 30 mM HEPES-BTP, pH 6.8, 3 mM  $MgSO_4$ , 3 mM ATP, 50 mM KCl, 0.5% ethanol plus or minus inhibitors or ionophores

Additions	ATPase activity $\mu$ mole $P_i$ /mg protein-h (%)
None	11.2(100)
Vanadate (200 $\mu$ M)	5.2(46.4)
Nitrate (50 mM)	7.8(69.6)
DCCD (10 $\mu$ M)	4.1(36.9)
CCCP (2 $\mu$ M)	16.0(142)
Gramicidin (5 $\mu$ g/ml)	15.8(141)

ATPase activity was inhibited by both vanadate and nitrate which suggested these microsomal vesicles were composed of at least two separate populations of membrane vesicles, presumably from the plasma membrane and the tonoplast (Marin, 1983; Sze, 1985). The majority of the ATPases in membrane vesicles from soybean roots were inhibited by vanadate (Table 1), an inhibitor of ATPase in the plasma membrane (Gallagher and Leonard, 1982). The concentration dependence of the ATPase activities on these two inhibitors are presented in Fig. 2. We have used 200  $\mu$ M of vanadate or 50 mM of nitrate in all other experiments with these two inhibitors.

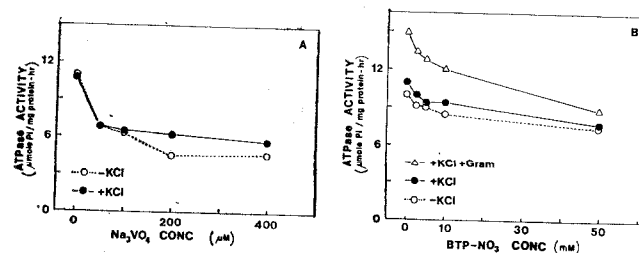


Fig. 2. Vanadate and nitrate inhibition of ATPase activity. Maximum inhibition of ATPase activity by vanadate was reached by 200  $\mu$ M. Concentration of KCl and gramicidin (Gram) was 50 mM and 5  $\mu$ g/ml, respectively.

*ATP-induced Formation of pH Gradient*

Quinacrine fluorescence quenching was used to monitor pH gradient formation in these microsomal vesicles. Percentage fluorescence is a direct measure of quinacrine uptake which depends on the formation of a pH gradient across the membrane (Lee *et al.*, 1982).

Quinacrine fluorescence quenching in these vesicles was induced by ATP (Fig. 3) and this suggested the formation of pH gradient (inside acid) across the membrane of vesicles. DCCD reversed the quenching probably by inhibition of ATPase activity (Table 1); reversal of quenching by CCCP, a protonophore, confirmed the ATP-induced formation of a pH gradient.

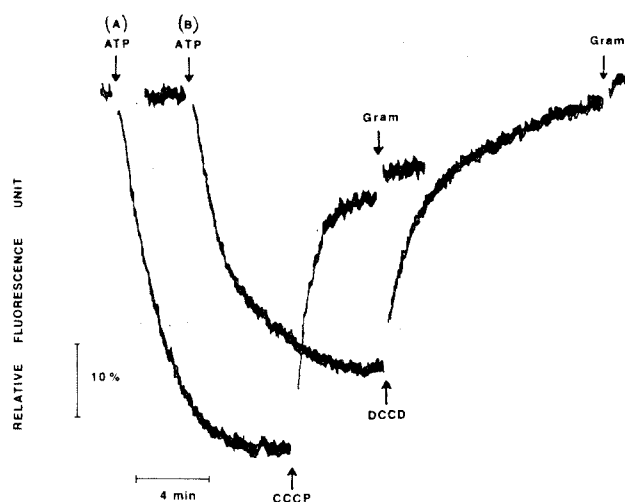


Fig. 3. ATP-induced quinacrine fluorescence quenching in microsomal vesicles from soybean roots. CCCP (5  $\mu$ M) reversed the quenching. ATP-dependent quenching was decreased by DCCD (50  $\mu$ M), an inhibitor of ATPase activity (see Table 1), Gram (5  $\mu$ g/ml gramicidin) completely reversed the quenching. Quinacrine fluorescence quenching was measured in a reaction mixture consisting of 10 mM HEPES-BTP, pH 7.4, 5 mM MgSO<sub>4</sub>, 50 mM BTP-Cl, 0.33 mM BTP-EGTA, 10  $\mu$ M quinacrine, 1.5 mM ATP and 0.13 mg/ml membrane vesicles. Excitation and emission wavelengths were 420 nm and 500 nm, respectively.

The ATP dependence was quite specific. Other substrates, such as ADP, AMP or GTP alone could not induce quinacrine fluorescence quenching in these vesicles (Fig. 4). If ADP or AMP was added after a steady state fluorescence quenching was induced by ATP, H<sup>+</sup> pumping was partially inhibited as quenching could be reversed. These results are consistent with the competitive inhibition of ADP or AMP with the substrate, ATP, as shown for the tonoplast-type H<sup>+</sup>-ATPase from oat roots (Wang and Sze, *in press*).

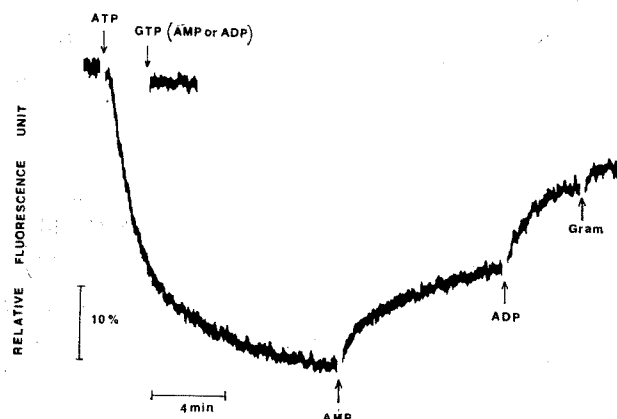


Fig. 4. Quinacrine fluorescence quenching was induced specifically by ATP. ADP, AMP or GTP alone at the same concentration of 1.5 mM had no effect on quinacrine fluorescence. ADP or AMP partially reversed ATP-dependent quenching.

As presented in Fig. 5, ATP-dependent pH gradient formation required  $Mg^{++}$  and was stimulated by BTP-Cl. BTP<sup>+</sup>, a large organic cation, presumably does not move across the membrane;  $Cl^-$ , a permeable anion, increased pH gradient formation and fluorescence quenching in part by dissipating the membrane potential (Churchill and Sze, 1984).

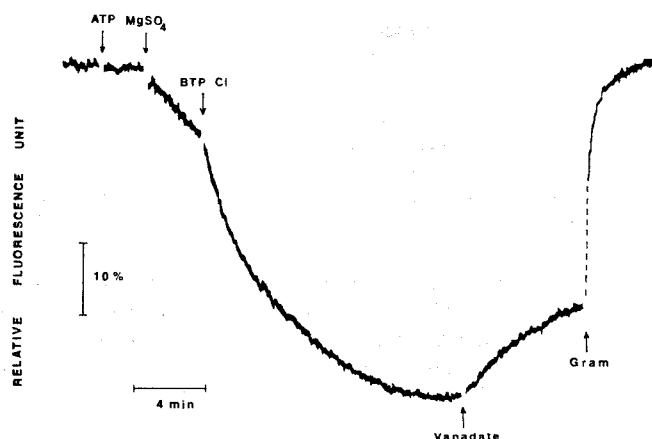


Fig. 5. ATP-induced quinacrine fluorescence quenching was increased by 50 mM BTP-Cl. Vanadate ( $200 \mu M$ ) partially decreased the quenching which was completely reversed by  $5 \mu g/ml$  of gramicidin.

#### *Vanadate and Nitrate Inhibition of pH Gradient Formation*

Similar to ATPase activity, quinacrine fluorescence quenching was inhibited

by both vanadate and nitrate. Vanadate at 200  $\mu$ M could inhibit as high as 50% of the total pH gradient generated (Fig. 6) when vanadate was added before or after addition of Mg-ATP. Since complete inhibition of ATPase activity was reached at a vanadate concentration of 200  $\mu$ M (Fig. 2), further inhibition of fluorescence quenching by 50 mM nitrate after the addition of 200  $\mu$ M vanadate (Fig. 7) suggested there were two types of H<sup>+</sup>-pumping ATPases, one was vanadate-sensitive and one was nitrate-sensitive.

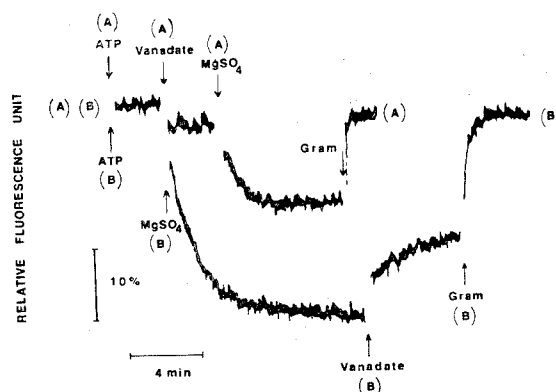


Fig. 6. Vanadate (200  $\mu$ M) added before (A) or after (B) Mg-ATP reduced quinacrine fluorescence quenching by 50%.

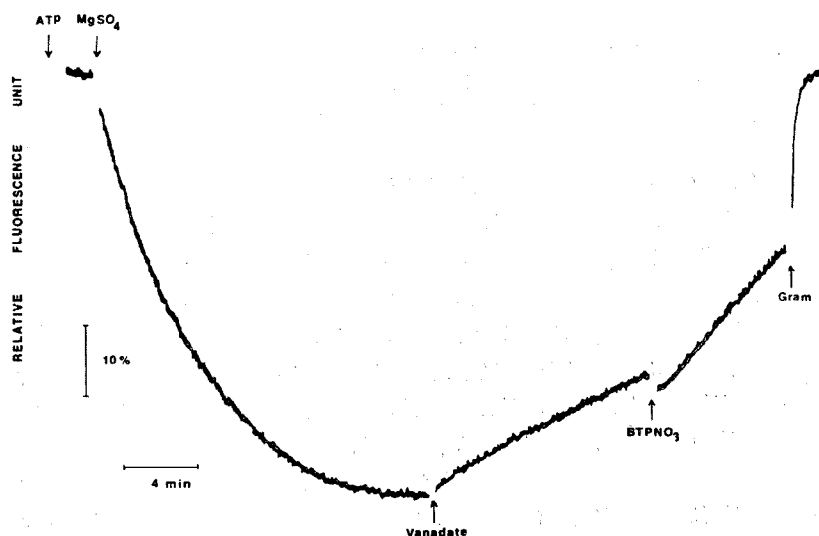


Fig. 7. ATP-induced quinacrine fluorescence quenching was partially inhibited by 200  $\mu$ M vanadate, BTP-NO<sub>3</sub> (50 mM) further decreased quenching. Gram = 5  $\mu$ g/ml gramicidin.

### *K<sup>+</sup> Decreased pH Gradient Formation*

If K-iminodiacetate (KIDA) was added after fluorescence quenching was induced by Mg-ATP, a decrease in quenching could be observed (Fig. 8) either when the vanadate-sensitive ATPase was inhibited with 200  $\mu$ M vanadate or when the nitrate-sensitive ATPase was eliminated with 50 mM BTP-NO<sub>3</sub>. Since K<sup>+</sup> does not inhibit the ATPase activity (Fig. 2) and IDA<sup>-</sup> is impermeant, the decrease in fluorescence quenching suggested K<sup>+</sup> collapsed the pH gradient by moving into the vesicles via a H<sup>+</sup>/K<sup>+</sup> exchange system. These results are similar to those described for microsomal vesicles from tobacco callus (Sze, 1983). However, pH collapse by K<sup>+</sup> in the presence of vanadate or nitrate suggests that a K<sup>+</sup>/H<sup>+</sup> antiport is associated with membranes of both types of H<sup>+</sup>-ATPases.

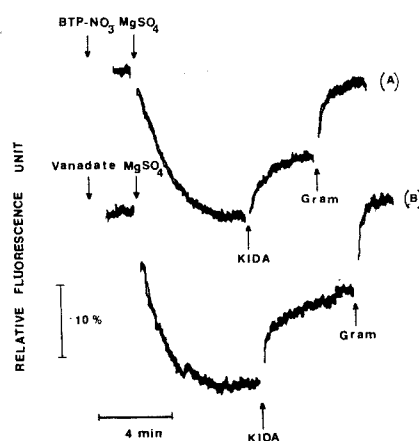


Fig. 8. Mg-ATP induced quinacrine fluorescence quenching was decreased by 50 mM of KIDA (K-iminodiacetate) in the presence of either 200  $\mu$ M vanadate or 50 mM of BTP-NO<sub>3</sub>.

### *Separation of Two Types of ATPase Activities*

Two pH optima could be distinguished when total Mg-ATPase activity was measured. By adding 200  $\mu$ M of vanadate, we obtained the activity of vanadate-insensitive ATPase with a single pH optimum of 8.1 and by subtracting the vanadate-insensitive ATPase activity from the total ATPase activities, we obtained the activity of vanadate-sensitive ATPase with a single pH optimum of 6.3 (Fig. 9). The nitrate-sensitive ATPase activity (like the vanadate-insensitive ATPase) had the same pH optimum of 8.1 (data not shown).

A linear sucrose gradient was used to separate the microsomal vesicles (Fig. 10). ATPase activity sensitive to vanadate was enriched at a buoyant density of 1.16–1.18 g/ml; activity sensitive to nitrate was greatest at 1.10–1.12 g/ml. Results from both the pH-dependence and sucrose-density gradient centrifugation studies



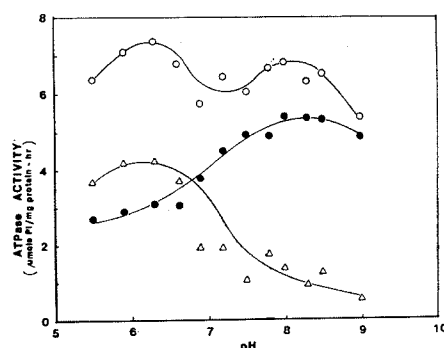


Fig. 9. Effect of pH on ATPase activities. In the absence of vanadate, ATPase activity (○) showed two pH optima. The vanadate-sensitive ATPase activity (Δ) had a pH optimum of 6.3; the vanadate-insensitive ATPase activity (●) had a pH optimum of 8.1. Vanadate-sensitive ATPase activities were obtained from the difference between ATPase activities in the presence and absence of 200  $\mu$ M vanadate. Vanadate-insensitive ATPase activities were determined in the presence of 200  $\mu$ M vanadate. Sodium azide at 0.2 mM and 0.1 mM ammonium molybdate were included in all reaction mixtures to eliminate mitochondria ATPase and unspecific phosphatases, respectively.

suggested the presence of two distinct types of ATPases in two different populations of sealed microsomal vesicles.

### Discussion

We have found at least two types of electrogenic H<sup>+</sup>-pumping ATPases associated with microsomal vesicles isolated from 2-day-old soybean roots. One type is probably located on the plasma membrane, a second type on the tonoplast.

Electrogenicity of these ATPases was concluded from ATP induced SCN<sup>-</sup> uptake into microsomal vesicles (Fig. 1). Evidence supporting the H<sup>+</sup>-pumping capabilities of these ATPases are: (1) SCN<sup>-</sup> uptake could be decreased by protonophore CCCP (Fig. 1); (2) ATPase activities were stimulated by CCCP (Table 1) and (3) quinacrine fluorescence quenching could be induced specifically by ATP (Fig. 4) but reversed by DCCD, an ATPase inhibitor, and CCCP (Fig. 3).

The conclusion that there are at least two types of H<sup>+</sup>-pumping ATPases in these vesicles and their probable association with plasma membrane and tonoplast was based on the following observations: (1) ATPase activity was inhibited by either vanadate or nitrate (Table 1; Fig. 2) which inhibits ATPases from plasma membrane and tonoplast, respectively. (2) ATPase activity sensitive to vanadate was greatest on membranes of 1.16-1.18 g/ml (Fig. 10), a density associated with most other plasma membranes; sensitivity to nitrate was greatest at 1.10-1.12 g/ml,

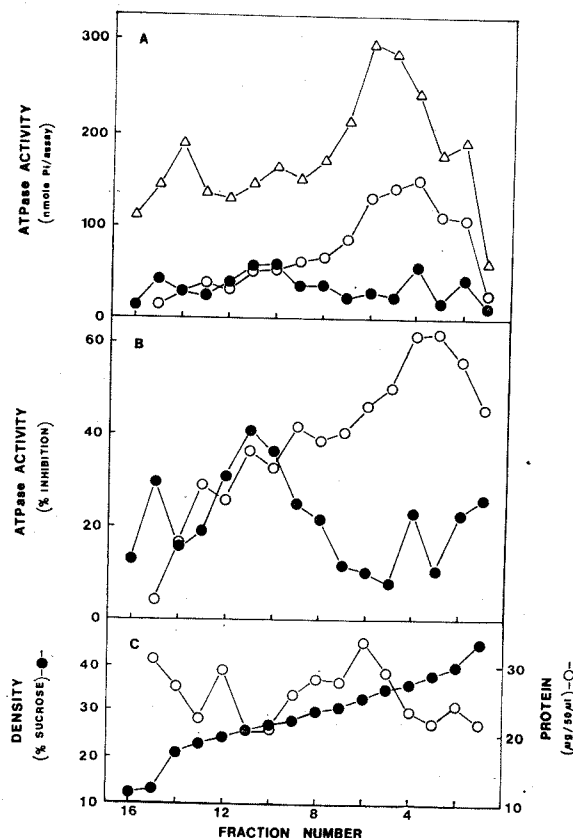


Fig. 10. Distribution of  $\text{NO}_3$ -sensitive or vanadate-sensitive ATPase in microsomal vesicles separated by a 20-40% linear sucrose gradient.

- A. Distribution of nitrate or vanadate inhibition of ATPase activity. ATPase activity was measured in the presence of 50 mM BTP- $\text{NO}_3$  or 200  $\mu\text{M}$  vanadate. Vanadate-sensitive ATPase (determined as in Fig. 9) was enriched at 31-40% sucrose. Nitrate-sensitive ATPase activities were obtained by subtracting ATPase activities in the presence of 50 mM BTP- $\text{NO}_3$  from the total ATPase activities. Sodium azide at 0.2 mM and 0.1 mM ammonium molybdate were included in all the reaction mixtures.
- B. Percentage of ATPase activity sensitive to vanadate (○) or nitrate (●). Percent sensitivity to an inhibitor was obtained by dividing the inhibitor-sensitive over the total ATPase activities. ATPase activity associated with membranes of 1.16-1.18 g/ml was 60% or more inhibited by 200  $\mu\text{M}$  vanadate. ATPase activity sensitive to nitrate was greatest on membrane with a buoyant density of 1.10-1.12 g/ml.
- C. Protein distribution and sucrose concentration fractionated from a linear sucrose gradient.

a density reported for tonoplasts. (3) Vanadate-sensitive ATPase in these vesicles had a pH optimum of 6.3, close to the value reported for plasma membrane ATPase; vanadate-insensitive ATPase had an optimum of 8.1, a value found several most ATPases of tonoplast origin. Our results are generally consistent with the properties of the plasma membrane and tonoplast H<sup>+</sup>-ATPases isolated from several other plant tissues (Sze, 1985; Churchill *et al.*, 1983; Bennett *et al.*, 1984; Hager and Biber, 1984; Rasi-Caldogno *et al.*, 1985; Mandala *et al.*, 1982; Poole *et al.*, 1984). Our conclusion on the location and inhibitor-sensitivity of these two types of H<sup>+</sup>-pumping ATPases generally agrees with the observation of Lew and Spanswick (1984b).

In some of our experiments, inhibition of ATP-induced fluorescence quenching by vanadate was less pronounced than inhibition by nitrate (Fig. 6), contrary to what should be expected from the greater magnitude of inhibition by vanadate on ATPase activity (Table 1; Fig. 2). Our tentative explanation is that the plasma membrane vesicles were more leaky to H<sup>+</sup> than those from tonoplast.

K<sup>+</sup> decreased ATP-induced quinacrine fluorescence quenching (Fig. 8) in 95% of our experiments. Na-IDA at 50 mM also decreased fluorescent quenching with similar magnitude (data not shown). Since K<sup>+</sup> decreased ATP-induced fluorescent quenching in the presence of either 200  $\mu$ M vanadate or 50 mM nitrate (Fig. 8), H<sup>+</sup>/K<sup>+</sup> exchange transport appeared to occur in vesicles from both the plasma membrane and the tonoplast. According to the working model, ATP-dependent H<sup>+</sup> pumping would originate from inside-out vesicles of the plasma membrane and right-side-out vesicles of the tonoplast (Sze, 1985). If so, the two H<sup>+</sup>/K<sup>+</sup> antiport systems would mediate removal of K<sup>+</sup> from the cytoplasm by transport out of the cell across the plasma membrane and across the tonoplast into the vacuole.

### Acknowledgements

We thank Ms. Yanzhi Wang for technical assistance. The Turner spectrofluorometer was generously provided by Dr. W. Vanderwoude (USDA, Beltsville). This study was supported in part by National Sciences Foundation Grant (USA; PCM 83-10928) to H. Sze and National Science Council Grant (ROC) to T.-H. Chen.

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## 由大豆根尖細胞所分離出膜囊上的兩種 Electrogenic, H<sup>+</sup>-Pumping ATPases

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由大豆根尖細胞所分離出之膜囊為測試 electrogenic H<sup>+</sup>-pumping ATPase 的材料。由 ATP 所引起的 S<sup>14</sup>CN<sup>-</sup> 進入膜囊作用為後加入之 CCCP 所逆轉顯示 ATP 所帶動的 electrogenic H<sup>+</sup> 泵使膜囊產生正值電位。MgATP 引起 quinacrine 螢光之剋減，Cl<sup>-</sup> 有進一步刺激剋減的作用。因 CCCP 或 gramicidin 均可逆轉 ATP 所引起之 quinacrine 螢光剋減，證實在膜囊內外已產生 ATP 所帶動之 pH 梯度分佈（內部較為酸性）。ADP 或 AMP 可部份抑制 ATP 所造成之 pH 梯度分佈，DCCD 則可完全抑制。所產生之 pH 梯度分佈約有 30-50% 受鈣酸根所抑制，其餘部份受硝酸根抑制。受鈣酸根所抑制的 ATPase 主要聚於 1.16-1.18 g/ml，最適 pH 為 6.3；受硝酸根所抑制之 ATPase 聚於 1.10-1.12 g/ml，最適 pH 為 8.1。這些特性顯示受鈣酸根所抑制之 H<sup>+</sup>-pumping ATPase 位於細胞膜；受硝酸根所抑制者位於液泡膜。K<sup>+</sup> 可使兩種 H<sup>+</sup> 泵所產生之 pH 梯度分佈消散，顯示在細胞膜及液泡膜上皆有 H<sup>+</sup>/K<sup>+</sup> 互換系統存在。